

Wave-Space Representation of the Electron Entropy in the Tight-Binding Approximation

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Abstract

The wave-space expression for the electron entropy averaged with respect to the hard-sphere reference system is obtained in the framework of the tight-binding model.

Keywords: Tight-binding model, electron entropy, wave space

The electron contribution to the entropy of a metal is expressed as follows (hereafter, in atomic units per atom):

$$S_e = (\pi k_b)^2 T n(\varepsilon_F) / 3, \quad (1)$$

where ε_F is the Fermi energy, k_b - Boltzmann constant, T - temperature, $n(\varepsilon)$ - density of electron states.

The average of S_e with respect to the hard-sphere (HS) reference system is

$$\langle S_e \rangle_{HS} = (\pi k_b)^2 T \langle n(\varepsilon_F) \rangle_{HS} / 3. \quad (2)$$

In the framework of the tight-binding model [1, 2]

$$\langle n(\varepsilon_F) \rangle_{HS} = 10 \sqrt{\frac{1}{2\pi\mu_{HS}}} e^{-\frac{\varepsilon_F^2}{2\mu_{HS}}}, \quad (3)$$

where

$$\mu_{HS} = 4\pi\rho \int_0^\infty \beta^2(r) g_{HS}(r) r^2 dr. \quad (4)$$

Here, $g(r)$ is the radial distribution function, ρ - mean atomic density,

$$\beta(r) = B \exp(-br), \quad (5)$$

where B and b are parameters.

In [2] was found that the magnitude of $e^{-\frac{\varepsilon_F^2}{2\langle\mu_m\rangle_{HS}}}$ is a constant for each metal since it depends on the number of d electrons per atom, N_d , only. As a result, Eq. (3) can be rewritten by the following way:

$$\langle n(\varepsilon_F) \rangle_{HS} = 10 \sqrt{\frac{1}{2\pi\mu_{HS}}} f(N_d) \quad (6)$$

To convert Eq. (2) to the wave space we use the relation

$$g(r) = 1 + \frac{1}{2\rho\pi^2} \int_0^\infty [S(q) - 1] \frac{\sin(qr)}{qr} q^2 dq, \quad (7)$$

where $S(q)$ is the structure factor.

After Fourier transform we obtained the following expression:

$$\langle S_e \rangle_{HS} = \frac{10}{\sqrt{2\pi}} f(N_d) \frac{(\pi k_b)^2 T}{3\sqrt{\mu_{HS}}}, \quad (8)$$

where

$$\mu_{HS} = \frac{\pi\rho B^2}{b^3} + \frac{8bB^2}{\pi} \int_0^\infty (S_{HS}(q) - 1) \left[\frac{q}{(4b^2 + q^2)} \right]^2 dq. \quad (9)$$

References

- [1] F.Ducastelle, J. Physique, 31 (1970), 1055.
- [2] F.Aryasetiawan, M.Silbert, M.J.Stott, Thermodynamic properties of liquid transition metals, J. Phys. F: Met. Phys., 16 (1986), 1419-1428.

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